

Physical Processes in hot gas

- Collisional ionization §13.4
- Photo-ionization §13.1

- Radiative Recombination §14
- Die-electronic Recombination
- Charge Exchange

- Collisional Ionization Equilibrium §14.9

- Radiative Cooling §34
- Thermal Conduction

□ When $T \gtrsim 10^4$ K, hydrogen starts to ionize and becomes dominant source of electrons

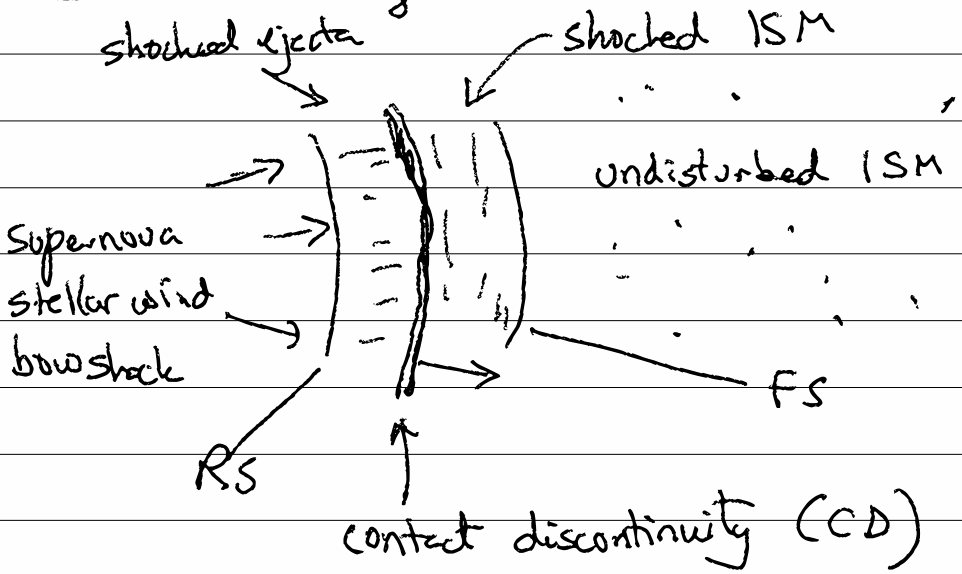
⇒ electron ionization

cooling by electron collisions

□ The ionization fraction depends only on temperature (not density)

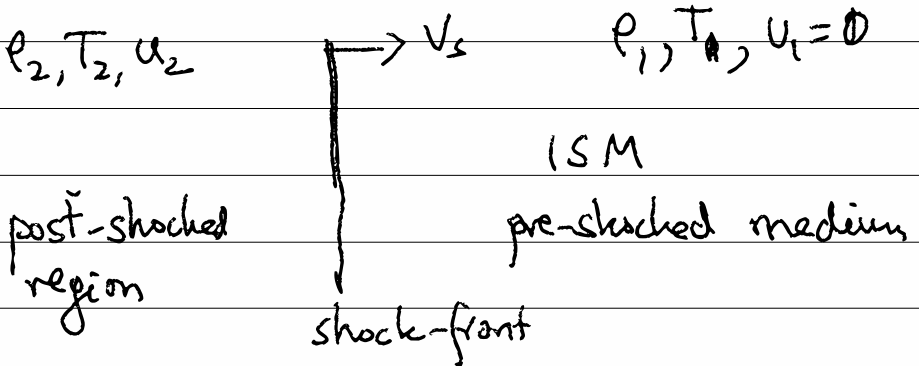
$$n_e n(X^{+n}) \sigma_{ci} = n_e n(X^{+(n+1)}) \alpha$$

Primer on strong shocks



FS: forward shock
 RS: reverse shock

Strong shock (summary, for now)



§ 32

$$\frac{p_2}{p_1} = \frac{\gamma+1}{\gamma-1} = 4 \quad \text{for } \gamma = 5/3$$

$$u_2 = \frac{\gamma-1}{\gamma+1} v_s = \frac{1}{4} v_s \quad \text{for } \gamma = 5/3$$

$$T_2 = \frac{2(\gamma-1)}{(\gamma+1)^2} \frac{\mu v_s^2}{k} = \frac{3}{16} \frac{\mu v_s^2}{k} \quad \text{for } \gamma = 5/3$$

Note: the shock accelerates particles $u_2 = \frac{1}{4} v_s$ and provides random velocity

The post-shocked particles gain velocity.

⇒ protons carry 2000x relative to electrons.

Note: We have ignored non-thermal processes

- acceleration of particle to relativistic energy
- amplification of magnetic field to levels comparable to thermal energy

Collisional Ionization by Electrons

For moderate energies $I < E \lesssim 3I$
 $I =$ ionization potential

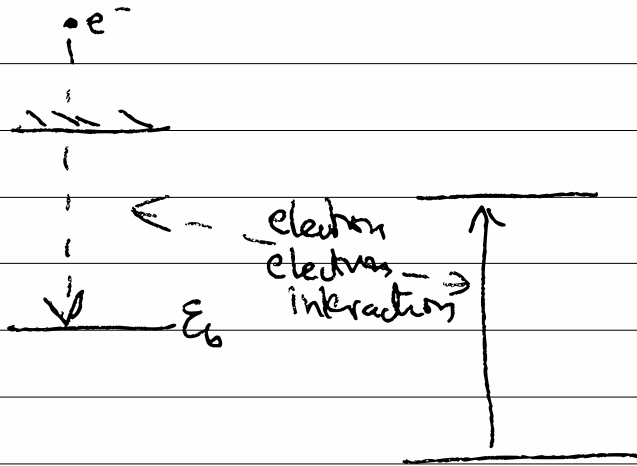
$$\sigma_{ci}(E) \approx C \pi a_0^2 \left(1 - \frac{I}{E}\right)$$

$$\begin{aligned} R_{ci} &= \int_I^{\infty} \sigma_{ci}(E) v f_E dE \\ &= C \pi a_0^2 \left(\frac{8kT}{\pi m_e}\right)^{1/2} e^{-I/kT} \end{aligned}$$

For H, e^- $C \approx 1.07$, $I = 13.6 \text{ eV}$

Dielectronic Recombination:

Radiative recombination is a slow process.
This process starts to dominate at high temperatures

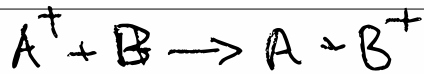


followed by radiative decay.

Example of a low-temperature dielectronic recomb.

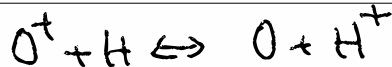
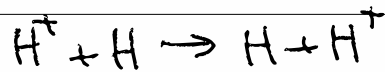
Electron recombines to fine-structure line of C^+
Another excited atom is raised to Rydberg state.

Charge Exchange



This is particularly effective when the reaction is exothermic or endothermic (but within kT).

Special Example:



Collision Ionization Equilibrium

Assume no incident photo-ionizing flux
optically thin to cooling radiation

$$n_e \langle \sigma v \rangle_{ci} n(X^{n-}) = n_e \langle \sigma v \rangle_{rr} n(X^{n+1})$$

ci ... collisional ionization

rr ... radiative recombination

$$\sigma_{rr}(E) = \frac{\partial}{\partial u} \frac{(1+E)^2}{Em_e c^2} \sigma_{pi}(h\nu = I+E)$$

↑ photo-ionization

However,

$$h \int \sigma_{pi}(\nu) d\nu = h \left(\frac{\pi e^2}{m_e c} \right) f_{pi}$$

but $\sigma_{pi}(\nu) = \sigma_{pi,0} \left(\frac{\nu}{\nu_0} \right)^{-3}$ ↑ oscillator strength

$$\therefore \sigma_{pi,0} = \frac{2\pi e^2}{m_e c} f_{pi} \frac{h}{I}$$

Thus

$$\frac{\langle \sigma v \rangle_{rr}}{\langle \sigma v \rangle_{ci}} = 2\pi \alpha^3 \frac{f_{pi}}{C} \frac{I}{kT} e^{I/kT}$$

When this ratio is unity $n(X^n) = n(X^{n+1})$

$$\frac{I}{kT} e^{I/kT} = \frac{C}{2\pi f_{pi}} \frac{1}{\alpha^3}$$

ex. $C=1$, $f_{pi} \approx 1$, $\frac{I}{kT} \approx 10.6$

Thus 50% ionization at $kT = \frac{I}{10}$
for H... $T \approx 15,000 \text{ K}$.

Including dielectronic recombination.

$$n_e \langle \sigma v \rangle_{ci} n(X^{+n}) = \left[\langle \sigma v \rangle_{rr} + \langle \sigma v \rangle_{de} \right] n_e n X^{+n+1}$$